

Chemistry 4000/6000 Fundamentals of Chemical Analysis - CTW (Lecture)
Syllabus (Spring 2014)

Lecture Instructor:

Dr. Bin Xu

Office: Courtland North 203

Phone: (404) 413-5567 Email: bxu6@gsu.edu

Office hours: T/Th. 1:30 PM – 2:30 PM or by appointment.

Meeting Time: Tuesday and Thursday, 11:00 – 11:50 AM

Location: Petit Science Center 362

Laboratory Instructor (all lab sessions start from the 1st week):

Monday Section: Dr. Gangli Wang 1:00 - 4:15 PM (Kell 698)

Tuesday Section: Dr. Cho-Ngi Ho 1:30 - 4:45 PM (Kell 698)

Wednesday Section: Dr. Cho-Ngi Ho 1:00 - 4:15 PM (Kell 698)

Tutorial (CHEM 4001) Instructor (Tutorial class starts from the 2nd week):

Jonathan Padelford

Time: T/Th 9:30 – 10:20 AM

Location: Langdale Hall 217

Email: jpadelford1@student.gsu.edu

Textbook: Quantitative Chemical Analysis, 8th edition, by Daniel C. Harris, W. H. Freeman and Co.

Laboratory manual: distributed on the first day of the lab.

Writing Guide: The ACS Style Guide (Effective Communication of Scientific Information), available on Reserve in the library.

Required Materials: safety glass; a bound laboratory notebook

Course Description:

Chemistry 4000/6000 is one of the WAC (Writing Across the Curriculum) and CTW (Critical Thinking through Writing) courses offered by the department of Chemistry of Georgia State University. The class includes two independent portions: lectures and laboratories. The lecture section of the course counts for 40% of the final grade; 60 % of the final grade will be derived from laboratory reports.

The units of the lectures are listed in the table on Page 3. Attendance will not be taken in the lectures. However, the students are strongly encouraged to attend the lectures. Hand-outs will be given during some lectures if necessary. There will be three in-class examinations (5% each) and one final examination (25%). Each exam will have two sections. A closed book section will include the most important principles or definitions from the units covered. In general those will be principles or theories that all chemists should know. In the open book section you will be asked to prove your understanding of

the principles and theories covered in the lectures. The problems in this section are generally longer and will demand students to “think” rather than “regurgitate” material.

The tutorial course (CHEM 4001) will focus on two parts: problem solving related to the homework assignments in the lecture; and all laboratory aspects. The homework practice will only be performed in the tutorial section, not in the lectures. Those practices, together with the in-class exams, are important to the preparation of the lab reports.

The course is designed to teach critical thinking and scientific writing following the American Chemical Society style. In the laboratory sections, the students will independently design experimental procedures, operate instruments commonly used in chemistry research, collect and analyze data, and draw conclusions and solve problems with scientific rationale. Four papers will be written using real data obtained in the laboratory section of the course. The papers will be returned with comments by lab instructors. The students will be allowed to rewrite, revise and resubmit. This process will continue until the student is satisfied with the grade. By the end of the semester the student should be able to demonstrate skills learned in introductory composition courses (proper grammar, sentence structure, etc.), demonstrate proper ACS style (the proper sections, properly cited references, etc.), and combine theory and real data in a logical manner (your explanation of data). The papers and laboratory notebook layout count 60% of the course grade.

Students are responsible for class preparation and for any material presented in class whether it is in the textbook or not. Chemistry is a highly structured course, with each new topic based on others previously developed. Thus it is critical for students to keep consistently up-to-date in their readings and assignments. To fall even one class period behind is to risk considerable difficulty in mastery of future material. Students should 1) review previous material, especially if not perfectly understood 2) complete reading assignments before the lecture in which the topics are covered, or at least immediately after the lecture and 3) complete assigned problems and exercises on time, with an emphasis on mastery of concepts and principles involved rather than looking up an answer and finding the formula that gives you that answer.

The syllabus provides a general plan for the course; deviations may be necessary.

University policy requires that faculty members must, on a date after the midpoint of the course to be set by the provost (or his designee) 1) give a WF to all those students who are on the roll but no longer taking the class and 2) report the last day that the student attended or turned in an assignment.

Policy Statement Regarding Student Integrity

The Georgia State University Policy on Academic Honesty is in force in this course, including but not necessarily limited to infractions in the areas of Plagiarism, Cheating on Examinations, Unauthorized Collaboration, Falsification, and Multiple Submissions. The university's policy is published in the *On Campus: The Student Handbook*, available to all members of the university community. Therefore, all tests taken must represent your individual unaided efforts. To receive or offer information during an examination is

cheating. The use of unauthorized supplementary materials during tests is also cheating. All laboratory work performed during the lab portion of a course must reflect your individual effort. Only original data obtained by your own in-lab experimentation are permitted to be used, except when specifically authorized by your laboratory professor. Data from supplementary sources (handbooks, reference literature, etc.) must be clearly referenced (title, author, volume, page(s), etc.). Falsification or destruction of data constitutes cheating. Conduct or actions that disrupt class or test periods or falsification of information related to chemistry courses by any student will be taken as violation of the policies of the Board of Regents of the University System of Georgia and the GSU Student Code of Conduct, Section 6.0. Any suspected offenses may be referred to the Department Chair or the Dean of Students for appropriate disciplinary action.

The last day to withdraw with a W is March 4th, 2014.

Final Exam: May 1st, 2014 (Thursday), 10:45 – 13:15, PSC 362

Due dates of papers: see the lab syllabus.

Grades (normalized):

95 -100: A+; 90-94: A; 87-89: A-; 85-86: B+; 80-84: B; 78-79: B-;
73-77: C+; 65-72: C; 60-64: C-; 57-59: D; <57:F

(To obtain a grade of C or above, the student must have the points from the lecture part no lower than 20.)

Dates	Major Contents (tentative)	
	Introduction, Lab, ACS style and writing	01/14, 01/16, 01/21, 01/23, 01/28, 01/30, 02/04 (review + titration)
	Error and Statistics	
	Statistics	
	Calibration Methods	
	Introduction to titration	
02/06	First exam	
	Equilibrium	02/11, 02/13, 02/18, 02/20, 02/25 (review)
	Activity, pH measurement, electric circuits	
	Systematic Treatment of Equilibrium	
02/27	Second exam	
	Acid Base monoprotic, Buffer	03/04, 03/06, 03/11, 03/13, 03/25, 03/27, 04/01(review)
	Acid Base polyprotic	
	EDTA	
04/03	Third exam	
	EDTA	04/08, 04/10, 04/15, 04/17, 04/22, 04/24 (review)
	Instrumental Methods	
	Electrochemistry: basic concepts	
* <i>Spring Break Week: Mar.17-21; No lecture and lab meeting</i>		
* <i>Final exam: May 1st (Thursday), 10:45 – 13:15, PSC 362</i>		