

## CHEM 2100 Course Syllabus Spring 2018

### Organic Chemistry Lab 1

Chemistry 2100

TIME: 4:00 – 8:45pm Tuesday

Room: 357 Petit Science Center

**Instructor:** Dr. Jeremiah D. Harden

301 Petit Science Center

**Office Hours:** Tues 12-3:30; Wed 8:30-11:30; Friday 8:30-11:30

E-mail: [jharden@gsu.edu](mailto:jharden@gsu.edu);

Send emails from **your GSU** email account **only** not through icollege, and the **COURSE TITLE MUST** and **YOUR NAME** be in the **SUBJECT** of the email

### Texts

(1) *Chem. 3100 Lab Manual* (available free during first lab)

### Optional Texts

(2) *Experimental Organic Chemistry* by Wilcox and Wilcox

(3) *Introduction to Spectroscopy* by Pavia, Lampman and Kriz

### GRADING

*Final Exam:	100 points
*Final Report	100 points
*Midterm Report	50 points
Quizzes, homework, *, **notebook, attendance, and preparation	<u>150 points</u>
	Total 400 points

\*Must be submitted to receive a passing grade.

\*\*Notebooks must be picked up within three weeks after final grade deadline (after which time they will be discarded)

A+: 96% A: 92%; A-: 89%; B+: 86% B: 82% B-: 78%, C+: 76% C: 72% C-: 68% etc.

**I DO NOT CURVE. I WILL ASIGN THE LETTER GRADE YOU EARN.**

**Tuesday Feb. 27<sup>th</sup>**

**Thursday April 20<sup>th</sup>**

**Tuesday April 24<sup>th</sup> by 6pm**

**Last day to withdraw with grade "W"**

**Final Exam during pre-lab lecture time in same room**

**Final report and Notebook are Due**

Please bring me a schedule of your **RELIGIOUS HOLIDAYS OBSERVANCE** by the **SECOND WEEK** of class. If you fail to do so you might miss important quizzes for this course.

### **Course Objective**

*In the **first part** of this project, you will isolate and purify compounds from natural products and you will practice liquid distillation and analysis of a known liquid sample. In the **second part** of the semester, three unknown liquids will be purified by distillation and the structure analyzed. Spectroscopy is an integral part of a modern organic chemistry laboratory. Therefore, you will have lectures throughout the semester on infrared spectroscopy (IR), and mass spectrometry (MS) where you will learn the fundamental principles behind each technique and how to interpret spectra in the assignment of organic structure. These spectroscopic techniques will be used during the laboratory portion of the course. . **You will be held responsible for the material discussed in lectures and that assigned from the textbooks in all quizzes and final exam.***

### **Part 1**

Weeks 1-5 Techniques for isolation and characterization of organic compounds  
Report sheets are to be filled out every week as the work is completed.

### **Part 2**

**Weeks 6-13 Term project; purification and identification of unknowns (Final report)**

**Final Report: neat liquid, low boiler and high boiler: A good boiling point is the most valuable information you will obtain from distillation.** Check what substances have the boiling points close to the one you have measured. You will find many different types of compounds with the same boiling point. Narrow down to those substances that contain the functional groups you tested in the lab. Then use the density and refractive index (RI) to assign the best candidate. If you have the MS, that gives you the molecular weight, draw structures and play with the fragmentation patterns to match the one given. When it comes the time for the identification of your unknowns **DO NOT LIMIT THE SEARCH TO THE DATABASE FOUND IN THE INTERNET. IT IS WRONG. SEARCH THE REFERENCE BOOKS** (in the lab, also available in the library).

***The final written examination will test your knowledge and comprehensive of basic techniques and processes employed in an organic laboratory and also the spectroscopic techniques used during the semester.***

### **WHAT IS IN YOUR FINAL?**

1. CONCEPTS AND CALCULATIONS USED IN FIRST 4 EXPERIMENTS, EXTRACTION, SUCH AS NEUTRALIZATION, DENSITY, SOLUBILITY(BENZOIC ACID/ ACETANILIDE), SUBLIMATION, RECRYSTALLIZATION, MELTING POINTS, ETC
2. ALL CHEMICAL TESTS TO DETERMINE FUNCTIONAL GROUPS OF ORGANIC COMPOUNDS ASSIGNED IN THE BOOK, NO ONLY THOSE THAT YOU'VE CARRIED OUT for your unknowns, ALL TAUGHT, you need to state the changes during the reactions (colors, heat, precipitates or any other observations, no only the name of the test)
3. DISTILLATION (SIMPLE AND FRACTIONAL, DIFFERENCES, equipments used)
4. GAS CHROMATOGRAPHY, CONCEPTS AND % COMPOSITION DETERMINATION OF A MIXTURE,
5. **IR interpretation is 35 % of the final**, PRACTICE WITH THE HOMEWORK AND QUIZZES
6. MASS SPECTROSCOPY, PRACTICE FRAGMENTATIONS

**Miscellaneous:**

1. Department of Chemistry Statement on Student Integrity applies to this course (see below).
2. Attendance to **lecture** and **lab** will be recorded. Absences can result in loss of points and lower grades (Sign-in/out of lab required).
3. Lab books must be recorded in ink at the time the measurements are made. **They will be to be graded during the lab section without announcing! Lab notebooks must be bound.**
4. **Safety glasses\* are required and must be worn at all times.** \*The student must bring a pair of safety glasses/goggles to the first lab. These may be purchased at the GSU Bookstore, the Georgia Bookstore, and most hardware stores. Students who are unable or forget to bring their glasses may **buy** a pair from their lab Coordinator by filling out a breakage form in the lab. Students who obtain glasses in this manner will pay for them at the time they check out of the lab. Students will not be allowed into the lab without their glasses/goggles
5. **Gloves MUST be worn when handling chemicals.**
6. **SAFETY, SAFETY, SAFETY, SAFETY, SAFETY. Failure to follow safety procedures will result in EXPULSION from that lab session with no make-up allowed and loss of credit. SAFETY, NOTHING GOES INTO THE SINK, USE THE HOODS!!! HOODS, HOODS, HOODS!!**

**DEPARTMENT OF CHEMISTRY POLICY STATEMENT REGARDING STUDENT INTEGRITY:**

The Department of Chemistry follows the university policy on academic honesty published in the "Faculty Affairs handbook" and the "On Campus: The Undergraduate Co-Curricular Affairs handbook." Any suspected offenses may be referred to the Department Chair for appropriate action.

All tests taken must represent your individual, unaided efforts. To receive or offer information during an examination is cheating. The use of unauthorized supplementary materials during tests is also cheating. All laboratory work performed during this course must reflect your individual effort. Only original data obtained by your own laboratory experimentation are to be used, except when specifically authorized by your laboratory professor. Data from supplementary sources (handbooks, reference literature, etc.) must be clearly referenced (title, author, volume, page(s), etc.). Falsification or destruction of data constitutes cheating.

*Very important:* **The following is a tentative schedule** of procedures and activities for **Chem. 3100 Summer 2012**. Changes and deviations from this syllabus will come and will be announced during class (quizzes, homework, and others). **SHOW UP ON TIME** to lectures to know what is going on.

Date	Week	<b><u>TENTATIVE</u> SCHEDULE OF LECTURES AND ACTIVITIES</b>	Miscellaneous reading assignments	Tentative Lab Manual Pages
January 9 <sup>th</sup>	1	Safety Test, Lab Check-in, and Students receive unknowns <b>Natural Product Extraction Trimyrustin from Nutmeg</b>	Lecture : Solid Liquid Extraction and Rotary Evaporation Notebooks	Write ALL unknown numbers in the notebook and roll  6, 12
January 16 <sup>th</sup>	2	<b>Natural Product Extraction</b> Caffeine from tea leaves.	Lecture: Aqueous Extraction IR Introduction	13, 14 handouts
January 23 <sup>rd</sup>	3	<b>Liquid/Liquid Extraction</b> Separation of Benzoic Acid and Acetanilide <b>Sublimation</b> of caffeine and infrared (IR) spectrum analysis. Introduction to Distillations.	Lecture: Multiple Extractions IR Continued Instrumentation Demo	15,16,18  47-66
January 30 <sup>th</sup>	4	<b>Esterification of Acetic Acid with an alcohol.</b> <b>IR collection</b>	Lecture: Esterification and Simple Distillation Analysis Discussion	17,18
February 6 <sup>th</sup>	5	<b>Breather Week</b> <b>Catch up Finish</b> <b>EVERYTHING for Midterm Report</b>	Papers due Next Week	
February 13 <sup>th</sup>	6	<b>Simple distillation:</b> Purification of neat liquid	<b>MIDTERM REPORT IS DUE TODAY.</b>	23
February 20 <sup>th</sup>	7	<b>Fractional distillation</b> Purification of an unknown mixture.	<b>MIDTERM REPORT IS DUE TODAY.</b>	24
March 27 <sup>th</sup>	8	<b>CONTINUE:</b> Separation of Low boiler (LB) and high boiler (HB), Boiling Point, <b>Gas Chromatography (GC)</b>	<b>GC ON</b>	27-29 4-68

March 6 <sup>th</sup>	9	Fractional distillation: <b>(IR) Spectroscopy, Chemical Tests</b> Continue separation of high boiler and low boiler <i>IMPORTANT: save LB and HB for chemical tests on week 10</i>	<b>GC ON</b>	4-68
March 13 <sup>th</sup>		<b>SPRING BREAK</b>		
March 20 <sup>th</sup>	11	Chemical Characterization Tests Introduction Mass Spectrometry	<b>GC ON</b>	4-68
March 27 <sup>th</sup>	12	Mass Spectrometry, <b>slides, request an mass spectrum of the unknown you have more difficulty to identify</b>	<b>Last week to use GC</b>	4-68
April 2 <sup>th</sup>	13	Miscellaneous Topics, How to study for the FINAL EXAM. Form of Final Report / Lab work completion, <b>only bp, chemical tests, IR, RI, density and books search are allowed, no more distillations</b>	<b>GC OFF</b> Last week to request MS of <b>ONLY ONE</b> of the unknowns	4-68
April 9 <sup>th</sup>	14	Miscellaneous Topics, How to study for the FINAL EXAM. Form of Final Report / Lab work completion, <b>only bp, chemical tests, IR, RI, density and books search are allowed, no more distillations</b>	<b>GC OFF</b> Last week to request MS of <b>ONLY ONE</b> of the unknowns	4-68
April 17 <sup>th</sup>	15	<b>FINAL EXAM</b>	<b>CHECK OUT</b>	
April 24 <sup>th</sup>		<b>Notebook &amp; Report Due in Box at the PSC 389 by 6 PM</b>	<b>REPORT IS NOT ACCEPTED WITHOUT NOTEBOOK</b>	