Inorganic Chemistry (Chem 4210/6210)

Spring 2019

Prerequisite: Chem 4120/6120

Professor: Dr. K.B. Grant, 423 NSC, (404) 413-5522, kbgrant@gsu.edu

Lecture: TTh 2:15 PM - 3:30 PM, Petit Science Center Room 233

Office Hours: TBA

Required Texts: "Inorganic Chemistry: Principles of Structure and Reactivity, <u>Fourth</u> <u>Edition</u>" James E. Huheey, Ellen A. Keiter, Richard L. Keiter (1993); "Descriptive Inorganic Chemistry, <u>Fifth Edition</u>" Geoff Rayner-Canham and Tina Overton (2010).

Optional Text: "Student Solutions Manual for Descriptive Inorganic Chemistry, 5th Edition" Geoff Rayner-Canham and Tina Overton.

Tentative Lecture Schedule: This schedule is a general guide and will be modified as needed.

<u>Date</u>	Chapter	<u>Topic</u> (Tu = Tuesday)
Jan 15 (Tu)	2 Huheey; 1 Rayner	Atomic Structure
Jan 17	2 Huheey; 2 Rayner	Atomic Structure & Periodicity
Jan 22 (Tu)	2 Huheey; 2 Rayner	Atomic Structure & Periodicity
Jan 24	4 Rayner	Metallic Bonding
Jan 29 (Tu)	4 Rayner	Metallic Bonding
Jan 31	4 Huheey; 5,6 Rayner	Ionic Bonding
Feb 5 (Tu)		Quiz 1
Feb 7	4 Huheey; 5,6 Rayner	Ionic Bonding
Feb 12 (Tu)	5 Huheey; 3 Rayner	Lewis Theory, VSEPR Theory
Feb 14	5 Huheey; 3 Rayner	Valence Bond Theory, Electronegativity
Feb 19 (Tu)	5 Huheey; 3 Rayner	Molecular Orbital Theory
Feb 21	5 Huheey; 3 Rayner	Molecular Orbital Theory
Feb 26 (Tu)	•	Exam 1
Feb 28	6 Huheey	Reactions of Covalent Inorganic Molecules
Mar 5 (Tu)	11 Huheey; 19 Rayner	Werner's Coordination Theory, Geometry
Mar 7	11 Huheey; 19 Rayner	Geometry, Isomerism
Mar 12 (Tu)	11 Huheey; 19 Rayner	Counting d Electrons, Nomenclature
Mar 14	11 Huheey; 19 Rayner	Thermodynamics, 18 e Rule, Valence Bond Model
Mar 19 (Tu)		No Class – Spring Break.
Mar 21		No Class – Spring Break.
Mar 26 (Tu)	11 Huheey; 19 Rayner	Crystal Field Theory
Mar 28	11 Huheey; 19 Rayner	Crystal Field Theory
April 2 (Tu)		Exam II
April 4	11, 12 Huheey	MO Theory of Metal Complexes; CN & Geometry

<u>Date</u>	<u>Chapter</u>	<u>Topic</u>	(Tu = Tuesday)
A 21.0 -	10.11.1	D 2 17 2 13 13 1	
April 9 (Tu)	13 Huheey	Reactions, Kinetics, and Mechanisms	
April 11	15 Huheey; 23 Rayner	Organometallic Chemistry	
April 16 (Tu)	15 Huheey; 23 Rayner	Organometallic Chemistry	
April 18	15 Huheey; 23 Rayner	Organometallic Chemistry	
April 23 (Tu)	19 Huheey; Class notes	Bioinorganic Chemistry	
April 25	19 Huheey; Class notes	Exam III	
May 2	Final Exam	<u>Cumulative</u> , 1:30 PM – 4:00 PM.	

<u>iCollege</u>: Please access iCollege for on-line course materials. For technical support, contact the IS&T Help Center at: <u>help@gsu.edu</u>, 404-413-HELP (4357), <u>www.gsu.edu/help</u>.

<u>Office Hours</u>: The Instructor will be available to meet with Students during scheduled office hours. Additional office hours will be arranged by appointment. Students are required to bring their notes. Walk-ins may not always be accepted.

Academic Honesty: The Department of Chemistry follows Georgia State University's Policy on Academic Honest. Students are expected to be familiar with and to comply with this policy. https://codeofconduct.gsu.edu/files/2018/07/2018-Here link Policy: to the 2019 Academic Honesty.pdf. All tests taken must represent your individual, unaided efforts. The following are examples of academic dishonesty: (i) to use an unauthorized homework key to complete a graded homework assignment; (ii) to sign an attendance sheet for a Student that is absent from class; (iii) to receive or offer information during an examination; (iv) to use unauthorized supplementary materials during tests; (v) to commit plagiarism on examinations and graded homework assignments (i.e., the act of presenting an individual's written work as one's own, without acknowledgment of the individual). Incidents related to academic honesty will be referred to the Chemistry Department Chair for appropriate action.

Grading: The grading scheme will be based on 500 points and will consist of three 100 point inclass exams, a 50 point in-class quiz, a cumulative final, homework quizzes (25 points), and attendance.

	Total:	500 points
	Final Exam	100
	Homework	25
	Attendance	25
	Quiz	50
	Exam III	100
	Exam II	100
Projected breakdown of points:	Exam I	100

Projected grade cut-offs:

A plus	96%
A	90%
A minus	88%
B plus	85%
В	75%
B minus	73%
C plus	70%
C	65%
C minus	63%
D	55%
F	less than 55%

Notes on Plus/Minus Grading: All Instructors have the option to award grades on a plus/minus scale. As per Departmental or College policy, Instructors decide on the criteria for the awarding of plus and minus grades. The following quality points are used to calculate GPAs.

4.30 A+: 4.00 A: 3.70 A-: B+: 3.30 B: 3.00 2.70 B-: C+: 2.30 C: 2.00 C-: 1.70 D: 1.00 F: 0.00 WF: 0.00

Evaluations: Student evaluations of the Instructor can be performed using the GoSOLAR/PAWS online evaluation system. Your constructive assessment of this course plays an indispensable role in shaping education at Georgia State. Upon completion of the course, please take time to fill out the online course evaluation.

Student Accommodations: Students who wish to request accommodation for a disability may do so by registering with the Office of Disability Services. Students may only be accommodated upon issuance by the Office of Disability Services of a signed accommodation plan and are responsible for providing a copy of that plan to Instructors of all classes in which an accommodation is sought.

<u>Miscellaneous</u>: Tuesday March 5th is the last day to withdraw from a class and receive a "W". Please note that any Student who enrolled in the course <u>without having completed</u> the required course prerequisites could be withdrawn from the course on this date <u>if your class average is a C minus or lower</u>. Any Students falling into this category should make arrangements to meet with the course Instructor on or before Tuesday March 5th.