

Chem 1212K Lab
Spring 2020
CRN 10158, CRN 13650 (Honors)

PreLab Lecture: Thursday 2:00 pm – 2:45 pm, **327 Classroom South**
Laboratory: Thursday 2:45 pm – 5:00 pm, **237 Classroom South**
(starts immediately after prelab)

Instructor: Dr. Gigi Ray
Courtland North 212
(404) 413-5540
gbray@gsu.edu

When sending emails, please use your official GSU email account and put “Chem 1212 Lab” in the subject line

Office Hours: Mondays 1:00 – 2:30pm, and Fridays 2:00 – 3:00pm, and by appointment.

Text: GSU Lab manual (to be handed out at first lab lecture; cost included in lab fees).
A course outline, schedule of activities (p.9), grading (p.7-8) is included in the lab manual.

Learning Outcomes:

Course will focus on reinforcing concepts and stoichiometric calculations from Chem 1211 and 1212 lectures. The students will synthesize a compound containing Cobalt, NH_3 and halide (Cl^- or Br^-). Then they will do QUANTITATIVE ANALYSIS to determine some properties of this compound, and will determine the formula of the synthesized product. This will involve precipitation reactions, acid-base titrations, redox titrations, and doing stoichiometric calculations to find the mass % of each element in the compound. Students will also learn how to prepare solutions of known concentration, and the proper handling and disposal of acids, bases and other common chemicals.

Lab Policies:

1. This lab is an individualized project-oriented lab, which includes synthesizing a compound, analyzing its composition, calculating and reporting results, identifying the compound, and explaining how the data obtained leads to the selected identity of the compound. Students will work independently in lab and report their own results.
2. Lab notebooks should be kept up to date. **Stitched, bound notebook required.** Leave 1 blank page at front of notebook for Table of Contents. Pages 7 and 37 of lab manual give details on notebook. ***All pages must be numbered and dated at top right. All data must be recorded in ink.*** Do not tear out pages, or whiteout/markout data. Instead present corrected data on another blank page. ***Notebooks must be signed by TA's or lab instructor when requested at beginning or end of every lab session.***
3. Quizzes will be given at the START of prelab, and will be closed book. No makeup quizzes given.
4. Take ALL notes in lab notebook during prelab lecture, including calculations done on board.
5. Before lab each week, students are responsible for READING and WRITING in their lab notebook about procedures for each new experiment in advance of lab (see schedule at end). CALCULATIONS on student's data needs to be done AT HOME IN NOTEBOOK, before next lab meeting.
A summary in your own words about the experiment including (i) purpose, (ii) materials, (iii) experimental methods (bullet points) must be written at HOME in your notebook before beginning an experiment. TAs will check notebooks before entering lab & students without written notes in their notebook, will NOT be allowed to enter the lab. During lab students are expected to record date in INK, DIRECTLY into the NOTEBOOK, not on other papers & later transferred to notebook.
6. **Safety glasses or goggles required to be worn at all times inside the lab.**

7. Dress appropriately. Students will not be allowed to enter the lab without appropriate clothing.
 - (i) No open-toed shoes (flip-flops, sandals, crocks, etc.)
 - (ii) No shorts or short skirts. Legs must be covered, no ankles showing (wear socks)
 - (iii) No sleeveless shirts/tops. Shoulders must be covered. Long sleeves advised. Use lab aprons.
8. **Students must turn in completed data sheets (in ink) with all calculations shown at the completion of each experiment** (p.41-51 of the lab manual, due dates below). **Points deducted for late data sheets, which will NOT be accepted after the second meeting following completion of an experiment.** Points will be deducted for late submissions, which are only allowed up to one week after the due date. Graded data sheets will NOT be returned to students (so keep a copy for yourself in your notebook), and late data sheets will not be accepted at the end of the semester. Typed final **data tables** and **sample calculations** must be included in the final lab report, including p.35 (summary data sheet). Preliminary report forms are found in the last pages of the lab manual, one for each experiment needing calculations.
9. **Calculations with correct significant figures and units must be included for full points on data sheets and final lab report. Complete all calculations on previous day's lab, before coming to next lab. Use lab time efficiently to complete experiments, not to finish previous calculations.**
10. **Cleaning up is part of the lab session. Students should stop working and begin cleaning up their work area, including their hood space, 20 minutes before the conclusion of the lab session.** Return all checked out items to TA at least 15 minutes before the end of lab (by 4:45pm). **Students must EXIT the lab by 5:00pm.** Points will be deducted if work area is left messy. Students need to put away all individual glassware & supplies in their locked drawer before leaving lab. Items left out on counters will be taken up and students will be charged for missing items at end of the semester.
11. **No food or drink allowed in the lab.** This includes chewing gum and candy. Put all food/drink away inside your book bag. Bags, coats, umbrellas, etc are to be stored in the compartments by the door; these items are not allowed on the floors, stools or lab counters. **Failure to follow safety rules will result in expulsion from the lab with no make-up allowed.**
12. **The preparation/handling of concentrated acid or ammonia solutions must be carried out under the hood. All experiments (except titrations) must be done in the hood.** Students must handle all chemicals safely and dispose of them in appropriate waste containers.
13. **No make-up labs allowed until after the semester mid-point on March 3, 2020.** Each student may only have one make-up lab session. Written authorization from lab instructor is needed before making up a lab. Students will only be admitted to make-up labs if the instructor determines that there is room in lab that day.

To pass the lab: students MUST (1) **take the written lab final exam**, (2) **turn in a final lab report**, and (3) **turn in their lab notebook**. These items are all **due at the beginning of pre-lab on Thursday April 23, 2020 (lab #13)**. It is important that students work on their lab report as experiments progress during the semester. Grading scheme is given on p. 7-8 of lab manual. Directions on writing a lab report are given on p.7 of lab manual and handout. *Note, the final lab report cannot be written in one night.*

First Session: Safety, check-in, crucibles weight experiment (pages 37 & 39 of the lab manual is taped into notebook as p. 3 & 4 of notebook. Use these as a model on how to write lab notebook. Every experiment needs to be dated (at top right of page). A similar format should be used for the rest of the experiments to be performed during the semester, including experiment title, date, purpose, materials, procedure, data (in table format), sample calculations, and conclusion.

Sessions 2 through 12: Preparation and determination of the formula for a cobalt-amino-halide complex, and determination of unknown concentrations and mass percentages of halide (Cl or Br), NH₃, Co.

Session 13 Final exam and lab check out. Lab notebook and final report (including the summary table on page 39 of the lab manual) are due Thursday April 23rd at start of prelab. Graded final exams, final reports, and lab notebooks can be viewed in the instructor's office in the first two weeks of the following semester, but will not be returned to the students. **No grades will be given via e-mail or by phone.**

TENTATIVE SCHEDULE OF EXPERIMENTS
Spring 2020 CRN 10158, CRN 13650 (Honors)

Thursdays 2:00pm – 5:00pm, Classroom South 327 & 237

Dr. Ray (gbray@gsu.edu)

Day	DATE	Session	CHEM 1212 LAB EXPERIMENTS
Thurs	Jan 23	1 SafetyQ	Check-in , Safety Video, Desks & Synthesis assigned. Review safety procedures and equipment use. Obtain constant weight of 2 Gooch crucibles (weigh on analytical balance, heat 30mins in oven, cool overnight in desiccator), do not touch with bare hands. Record Balance number. Review Buret & Analytical Balance use.
Thurs	Jan 30	2	Start Synthesis of Co-aquo-NH₃-halide (Cl⁻ or Br⁻) (assigned by instructor). Also reweigh crucible, heat, then cool in desiccator. Want constant weight (2 measurements) within +/- 0.0005 grams.
Thurs	Feb 6	3	Complete Synthesis of Co-aquo-NH₃-halide (Cl⁻ or Br⁻) . Also reweigh crucible, heat, cool & reweigh to constant weight +/- 0.0005 g.
Thurs	Feb 13	4	Prepare ~0.3M HCl solution. Titrate with primary standard THAM to standardize HCl . - Submit % yield & product color of synthesized compound.
Thurs	Feb 20	5	Precipitation of Halide . Determination of %Cl ⁻ or %Br ⁻ in synthesized compound. - Submit "Preliminary HCl Molarity" Data Sheet and calculations (p.43)
Thurs	Feb 27	6	Complete Precipitation of Halide . Determination of %Cl ⁻ or %Br ⁻ in synthesized compound.
Tues	Mar 3		Last day to withdraw (from BOTH lab and lecture)
Thurs	Mar 5	7	Distillation of NH₃ into boric acid solution in the hood then titration of NH ₃ with standardized HCl. (BOTH distillation & acid-base titration must be carried out in SAME lab session). Calculate %NH ₃ in synthesized compd. - Submit "Preliminary %Halide" Data Sheet (p.41). Show full calculations with correct significant figures.
Thurs	Mar 12	8	Continue distillation and titration of NH₃ 2 more times. Determine %NH₃
Thurs	Mar 19		No Lab, Spring Break week
Thurs	Mar 26	9	Spectrophotometric determination of % Co. ChemDraw Computer Activity: Acid-Base Reaction Mechanisms (Honors students and other interested students) - Submit "Preliminary Report on %NH₃" Data Sheet (p. 45)
Thurs	Apr 2	10	Preparation and standardization of Na₂S₂O₃ . Prepare Na ₂ S ₂ O ₃ and titrate with KIO ₃ to use for Redox Titration for Analysis of H ₂ O ₂ using Na ₂ S ₂ O ₃ . - Submit "Preliminary Report on %Co" Data Sheet (p. 47)
Thurs	Apr 9	11	Redox Titration for Analysis of H₂O₂ using Na₂S₂O₃ . Titrate H ₂ O ₂ with KIO ₃ and standardized Na ₂ S ₂ O ₃ to determine % H ₂ O ₂ . - Submit "Preliminary Report Standardization Na₂S₂O₃" Data (p. 49)
Thurs	Apr 16	12	Make-up Lab, Complete all experiments. Clean-up and Checkout. - Submit "Preliminary Report on %H₂O₂" Data Sheet (p. 51)
Thurs	Apr 23	13	Take Final Exam, Submit Lab Notebook and Final Lab Report (including summary data sheet p. 35). Submit Honors Activity. Clean-up & Lab Checkout.

Chemistry 1212 Lab Agreement

Spring 2020

CRN 10158, CRN 13650 (Honors)

Thursdays 2:00pm – 5:00pm, Classroom South 327 & 237

Dr. Ray (gbray@gsu.edu)

I have read the entire syllabus and understand the grading system and agree to abide by the course policies and all the additional information provided in it, including the following:

1. **No make-up tests or quizzes** will be given.
2. The instructor reserves the right to assign seating during exams and quizzes. Students need to show their GSU Panther I.D. card when taking exams and quizzes.
3. Cell-phone calculators and programmable calculators are not allowed. Cell-phones need to be kept either in purses or book-bags during exams or quizzes, and should be turned off at all times.
4. **Failure to follow safety procedures (including appropriate clothing & shoes)** will result in expulsion from the lab, with no opportunity to do makeup labs and with a loss of credit.
5. **Failure to come prepared to lab (with purpose, procedure, blank data tables in permanent lab notebook)** will result in the student sitting outside the lab, and writing this material in the notebook before being allowed to enter the lab.
6. **Preliminary reports on all calculations, in ink, must be submitted at the time of the completion of each experiment (due dates in syllabus). Points deducted for late data sheets, which will NOT be accepted after the second meeting following completion of an experiment.** A typed version of data tables must be included in the appendix section of the final report; otherwise, you will be assigned the minimum amount of points for those experiments not submitted (if you have supporting data in your lab notebook). Preliminary report forms are found in the last pages of the lab manual, one for each experiment needing calculations.
7. **For full points show sample calculations with correct significant figures and units. Complete all calculations on previous day's lab, before coming to next lab. Use lab time efficiently to complete experiments, not to finish previous calculations.**
8. **The preparation/handling of concentrated acid or ammonia solutions must be carried out under the hood.** Unused concentrated acid or ammonia solutions must be diluted by adding them to water, under the hood. The amount of water to be used in the dilution depends on the amount of reagent needed to be diluted so it will not fume (about 1 in 10 dilution). The diluted solutions will be discarded in the waste drum located in the lab. Glassware used for the preparation/handling of concentrated acid or ammonia solutions must be rinsed with enough water, under the hood, and the combined rinses must then be placed in the waste drum.
9. **The clock schedule for the lab must be followed.** Students are not to enter occupied labs before the scheduled start time of the lab itself. *For each scheduled lab section, clean-up must be completed by the end time of the lab (5:00 pm) so as not to disrupt the next class.*
10. **Make-up lab policies are posted** on the door outside the lab, and will only be allowed after Mar 3, 2020 ONLY if space available. A written authorization from your lab instructor is required.
11. To pass the lab students **MUST** (1) **take the written lab final exam**, (2) **turn in a final lab report**, and (3) **turn in their lab notebook**. These items are **due at the beginning of pre-lab on Thursday April 23, 2020** (during lab #13). Note, the final lab report cannot be written in one night.

Name (print in ink): _____

Signature (in ink): _____ Date: _____