DEPARTMENT OF CHEMISTRY
ORGANIC CHEMISTRY LAB I CHEM 3100 Spring 2014

Pre-Lab Lecture (mandatory): Tuesday 4:30 to 5:20 pm, PSC 362
Lab: (mandatory): Tuesday 5:30 to 9:15 pm, PSC 357

NOTE: You must attend lab lecture and understand the concepts before entering into the laboratory to perform any experiment.

TEXTBOOKS and LABORATORY MATERIALS

- Chem. 3100 Lab Manual (will be distributed during the first lab)
- Experimental Organic Chemistry by Wilcox and Wilcox, second edition, 1995 (Required first day of lab)
- Introduction to Spectroscopy by Pavia, Lampman and Kriz 4th edition (Optional)
- The Organic Chemistry Laboratory Survival Manual, Zubrick 8th edition (Optional)
- Stitched notebook (mandatory and required first day of lab)
- Safety glasses/goggles (mandatory and required first day of lab)

Also NOTE: Absolutely No shorts, No sleeveless, No open toe/open top shoes, No untied long hair, No Crocs, No contact lenses will be allowed in the lab. Safety will be enforced in the lab.

Instructor: Dr. Julianne Caton-Williams
Office Location: 210 Courtland North (Building 9 on the GSU map)
Email: jcatonwilliams1@gsu.edu
Office Hours: Monday 3:30 pm to 5:00 pm, and by appointment

Course Objective
The course has been divided into two parts. In the first part (Week 1-5), students will isolate and purify compounds from natural products, such as tea leaves and nutmeg to learn different extraction techniques such as liquid-liquid extraction and solid-liquid extraction. The products isolated will be purified by recrystallization, sublimation etc. Students will also have the opportunity to purify and characterize a “known” liquid by distillation for the purpose of calibration of equipment. This technique acquired will be applied to the second part of the lab. In the second part (Week 6-13), students will purify three unknown liquids by distillation. The distilled products will be analyzed using Gas-Liquid Chromatography (GC) to know the purity, and spectroscopic techniques such as infrared spectroscopy (IR), and mass spectrometry (MS) to assist in the identification of the structure. Students will learn the fundamental principles behind each technique and master how to interpret spectra in the assignment of their unknown organic structures. Also, students will learn how to write scientific laboratory reports (midterm and final) which would be graded. The midterm report includes only the first 4 experiments in the first part and will not be part of the final report.

Teaching Schedule
The lab/lecture schedule listed on page 7 of the GSU laboratory manual will be adhered to as far as possible. You would be informed of any changes.

Students’ preparedness
- For each lab day, we will have a discussion on the procedure (method) for the lab expected to perform on the lab day, so ensure to read the lab from the text and any handouts given prior to coming to the lab.
- Do not miss lectures otherwise you will not know what is going on.
You are responsible for the material discussed in lectures and assignments from the textbooks to apply in all quizzes and final exam.
You are to adhere to the lab safety rules presented and follow instructions carefully.
No use of cell phones during lecture and labs.

**Lab policy**
- Students who need to make-up a lab should obtain an authorization from the lab instructor before attending a different lab section.
- Cleaning up is part of the lab session. Students should stop working and begin cleaning up their work area, including their hood space, at least **25 minutes** before the conclusion of the lab session.

**Quizzes and homework**
- Quizzes and homework would be announced. Please check D2L for information pertaining to this course.

**GRADING SCHEME**

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<tr>
<th>Points</th>
<th>Percentage</th>
<th>Letter grade</th>
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<tbody>
<tr>
<td>≥360</td>
<td>≥ 95%</td>
<td>A’</td>
</tr>
<tr>
<td>360 to &lt; 380</td>
<td>90% to &lt; 95%</td>
<td>A</td>
</tr>
<tr>
<td>348 to &lt; 360</td>
<td>87% to &lt; 90%</td>
<td>A’</td>
</tr>
<tr>
<td>336 to &lt; 348</td>
<td>84% to &lt; 87%</td>
<td>B’</td>
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<tr>
<td>320 to &lt; 336</td>
<td>80% to &lt; 84%</td>
<td>B</td>
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<tr>
<td>312 to &lt; 320</td>
<td>78% to &lt; 80%</td>
<td>B</td>
</tr>
<tr>
<td>300 to &lt; 312</td>
<td>75% to &lt; 78%</td>
<td>C’</td>
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<tr>
<td>276 to &lt; 300</td>
<td>69% to &lt; 75%</td>
<td>C</td>
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<tr>
<td>260 to &lt; 276</td>
<td>65% to &lt; 69%</td>
<td>C’</td>
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<tr>
<td>≥ 240 to &lt; 260</td>
<td>60% to &lt; 65%</td>
<td>D</td>
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Letter grades are assigned based on the following scale (which may be varied slightly):

The grade letter you earn will be assigned. You will have access to most of your grades in Desire2Learn.

**Important Dates**

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<thead>
<tr>
<th>Date</th>
<th>Event</th>
<th>Details</th>
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<tbody>
<tr>
<td>January 14th</td>
<td>Labs begin, check in, safety quiz</td>
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<tr>
<td>January 20th</td>
<td>Holiday (MLK)</td>
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<td>March 4th</td>
<td>Last day to withdraw with grade “W”</td>
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<tr>
<td>March 17th to 23rd</td>
<td>Spring break</td>
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<tr>
<td>April 22nd</td>
<td>Last day for Tuesday Lab</td>
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<tr>
<td>April 22nd</td>
<td>Final exam (4:30 - 6:30 pm)/ check out</td>
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<tr>
<td>April 22nd</td>
<td>Final report and notebook due in MY OFFICE 210 Courtland North by 5:00 pm</td>
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The University requires that faculty members must, on a date after the mid-point of the course to be set by the Provost (or his designee)

1. Give a WF to all students who are on their rolls but are no longer taking the class
   and
2. report the last day the student attended or turned in an assignment.

Students who are withdrawn may petition the Departmental Chair for reinstatement into their classes.

NOTE:
Students are requested NOT to bring cellular telephones and/or pagers to lectures and labs, or exams. Persons violating this request will be asked to leave the room.

Miscellaneous:
1. Department of Chemistry Statement on Student Integrity applies to this course (see below).
2. Attendance to lecture and lab will be recorded. Absences can result in loss of points and lower grades (Sign-in/out of lab required).
3. Lab books must be recorded in ink at the time the measurements are made. They will be graded during the lab section without announcing! Lab notebooks must be bound.
4. Safety glasses* are required and must be worn at all times. *The student must bring a pair of safety glasses/goggles to the first lab. These may be purchased at the GSU Bookstore, the Georgia Bookstore, and most hardware stores. Students who are unable or forget to bring their glasses may buy a pair from their lab Coordinator by filling out a breakage form in the lab. Students who obtain glasses in this manner will pay for them at the time they check out of the lab. Students will not be allowed into the lab without their glasses/goggles or properly attired.
5. Gloves MUST be worn when handling chemicals.
6. SAFETY. Failure to follow safety procedures will result in EXPULSION from that lab session with no make-up allowed and loss of credit. SAFETY, NOTHING GOES INTO THE SINK, USE THE HOODS!!!

Please bring me a schedule of your RELIGIOUS HOLIDAYS OBSERVANCE the SECOND WEEK of class. If you fail to do so you might miss important quizzes for this course. Please check Desire2Learn regularly for announcements to include homework and quizzes.

DEPARTMENT OF CHEMISTRY POLICY STATEMENT REGARDING STUDENT INTEGRITY:
The Department of Chemistry follows the university policy on academic honesty published in the “Faculty Affairs handbook” and the “On Campus: The Undergraduate Co-Curricular Affairs handbook.” Any suspected offenses may be referred to the Department Chair for appropriate action.
All tests taken must represent your individual, unaided efforts. To receive or offer information during an examination is cheating. The use of unauthorized supplementary materials during tests is also cheating. All laboratory work performed during this course must reflect your individual effort. Only original data obtained by your own laboratory experimentation are to be used, except when specifically authorized by your laboratory professor. Data from supplementary sources (handbooks, reference literature, etc.) must be clearly referenced (title, author, volume, page(s), etc.). Falsification or destruction of data constitutes cheating.

Very important: The following is a tentative schedule of procedures and activities for Chem. 3100 Spring of 2014. Any changes and deviations from this syllabus will be announced during class (quizzes, homework, and others). Do not miss lectures otherwise you will not know what is going on.

Deviations from this syllabus may be required!